

# Standard states of the elements

The most stable form of an element  
at 298 K and 1 atm (“STP”)

## Metals

All solids except one  
(which one?)

## Metalloids

All solids

## Nonmetals

Atomic gases – Noble gases  
He, Ne, Ar, Kr, Xe, Rn

Diatomics – halogens and H<sub>2</sub>, N<sub>2</sub>, O<sub>2</sub>  
H<sub>2</sub>, N<sub>2</sub>, O<sub>2</sub> (gas)

F<sub>2</sub> (gas)  
Cl<sub>2</sub> (gas)  
Br<sub>2</sub> (liquid)  
I<sub>2</sub> (solid)

} Halogens (group 7)

Other nonmetals – solids  
C (graphite), S, P, Se

# PERIODIC TABLE of the ELEMENTS

Standard States

MAIN GROUPS

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1A 1 <b>H</b> 1.008		2A 2 <b>He</b> 4.003		TRANSITION METALS													
3 <b>Li</b> 6.941	4 <b>Be</b> 9.012	5B 5 <b>V</b>	6B 6 <b>Cr</b>	7B 7 <b>Mn</b>	8B 8 <b>Fe</b>	8B 9 <b>Co</b>	8B 10 <b>Ni</b>	1B 11 <b>Cu</b>	2B 12 <b>Zn</b>	31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>		
11 <b>Na</b> 22.990	12 <b>Mg</b> 24.305	3B 3 <b>Ti</b>	4B 4 <b>Sc</b>	5B 5 <b>Nb</b>	6B 6 <b>Mo</b>	7B 7 <b>Tc</b>	8B 8 <b>Ru</b>	8B 10 <b>Rh</b>	1B 11 <b>Pd</b>	2B 12 <b>Ag</b>	31 <b>Cd</b>	32 <b>In</b>	33 <b>Sn</b>	34 <b>Sb</b>	35 <b>Te</b>	36 <b>I</b>	
19 <b>K</b> 39.098	20 <b>Ca</b> 40.078	21 <b>Sc</b>	22 <b>Ti</b>	23 <b>V</b>	24 <b>Cr</b>	25 <b>Mn</b>	26 <b>Fe</b>	27 <b>Co</b>	28 <b>Ni</b>	29 <b>Cu</b>	30 <b>Zn</b>	31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>
37 <b>Rb</b> 85.468	38 <b>Sr</b> 87.62	39 <b>Y</b>	40 <b>Zr</b>	41 <b>Nb</b>	42 <b>Mo</b>	43 <b>Tc</b>	44 <b>Ru</b>	45 <b>Rh</b>	46 <b>Pd</b>	47 <b>Ag</b>	48 <b>Cd</b>	49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>	54 <b>Xe</b>
55 <b>Cs</b> 132.91	56 <b>Ba</b> 137.33	57 <b>La*</b>	72 <b>Hf</b>	73 <b>Ta</b>	74 <b>W</b>	75 <b>Re</b>	76 <b>Os</b>	77 <b>Ir</b>	78 <b>Pt</b>	79 <b>Au</b>	80 <b>Hg</b>	81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>	86 <b>Rn</b>
87 <b>Fr</b> [223]	88 <b>Ra</b> [226]	89 <b>Ac**</b> [227]	104 <b>Rf</b> [261]	105 <b>Db</b> [262]	106 <b>Sg</b> [266]	107 <b>Bh</b> [264]	108 <b>Hs</b> [265]	109 <b>Mt</b> [268]	110 [269]	111 [272]	112 [277]		114 [285]		116 [289]		118 [293]

\* LANTHANOIDS

\*\* ACTINOIDS

58 <b>Ce</b> 140.12	59 <b>Pr</b> 140.91	60 <b>Nd</b> 144.24	61 <b>Pm</b> [145]	62 <b>Sm</b> 150.36	63 <b>Eu</b> 151.96	64 <b>Gd</b> 157.25	65 <b>Tb</b> 158.92	66 <b>Dy</b> 162.50	67 <b>Ho</b> 164.93	68 <b>Er</b> 167.26	69 <b>Tm</b> 168.93	70 <b>Yb</b> 173.04	71 <b>Lu</b> 174.97
90 <b>Th</b> 232.04	91 <b>Pa</b> 231.04	92 <b>U</b> 238.03	93 <b>Np</b> [237]	94 <b>Pu</b> [244]	95 <b>Am</b> [243]	96 <b>Cm</b> [247]	97 <b>Bk</b> [247]	98 <b>Cf</b> [251]	99 <b>Es</b> [252]	100 <b>Fm</b> [257]	101 <b>Md</b> [258]	102 <b>No</b> [259]	103 <b>Lr</b> [262]