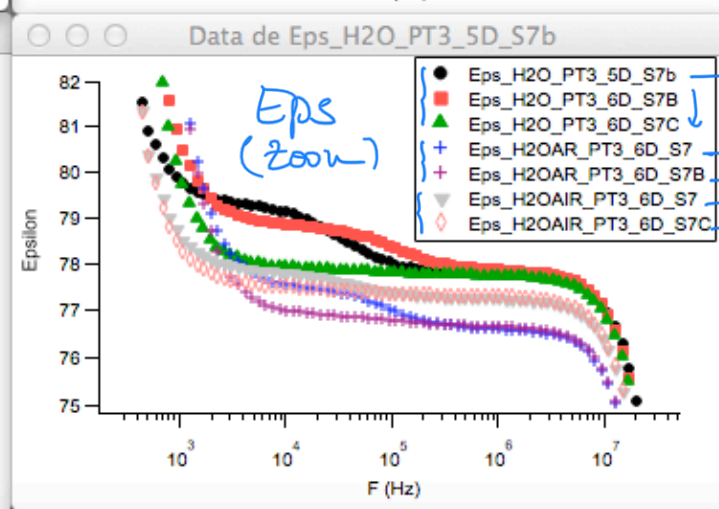
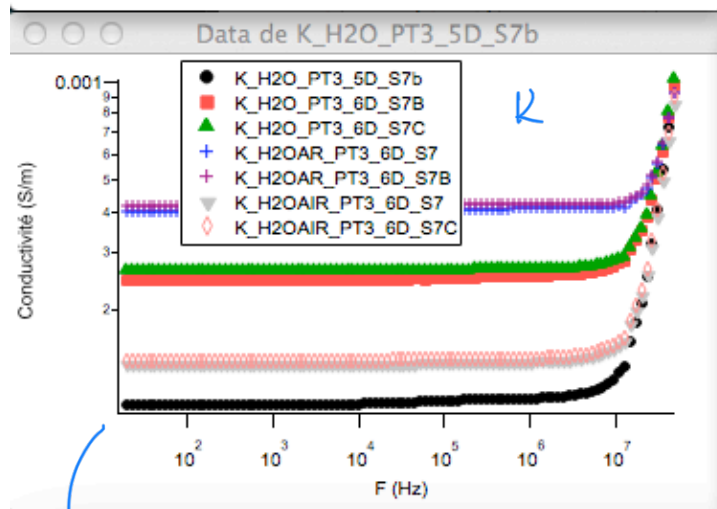
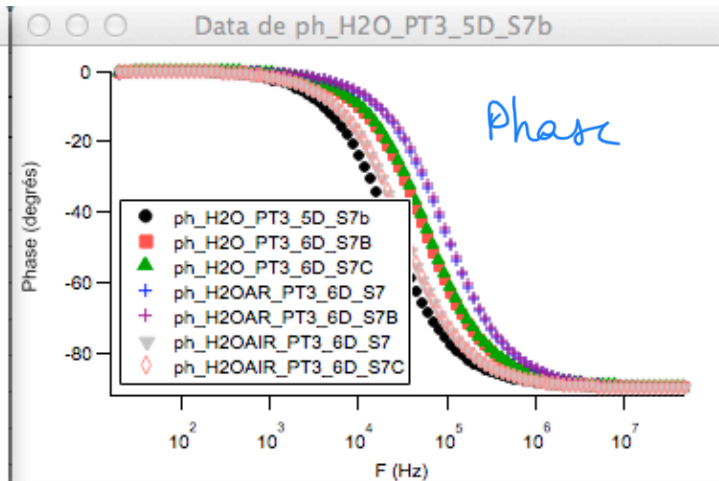
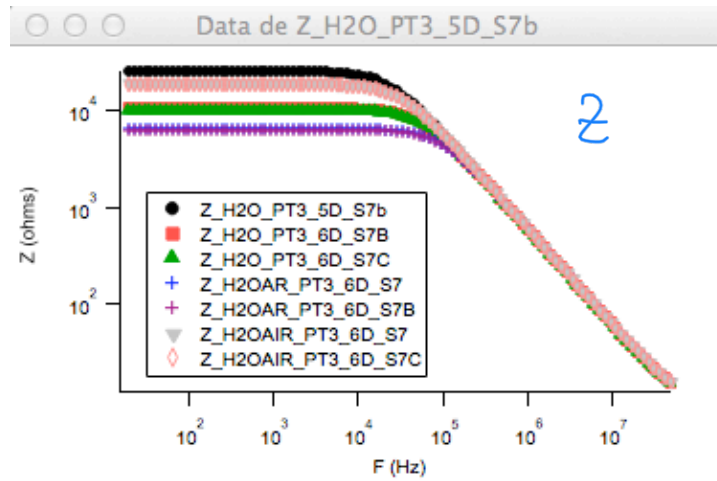


Tests on water - Cell 3 $D=6.1\mu m$

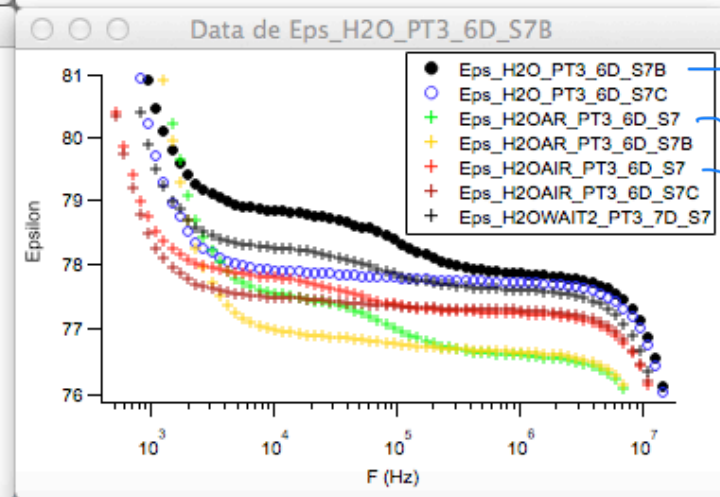
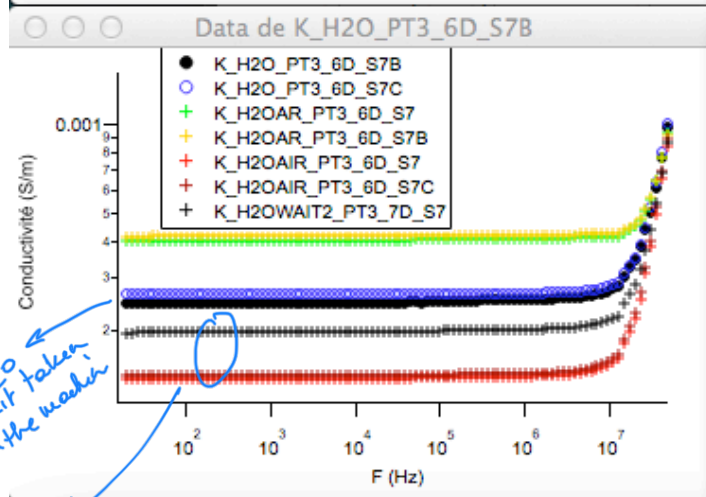
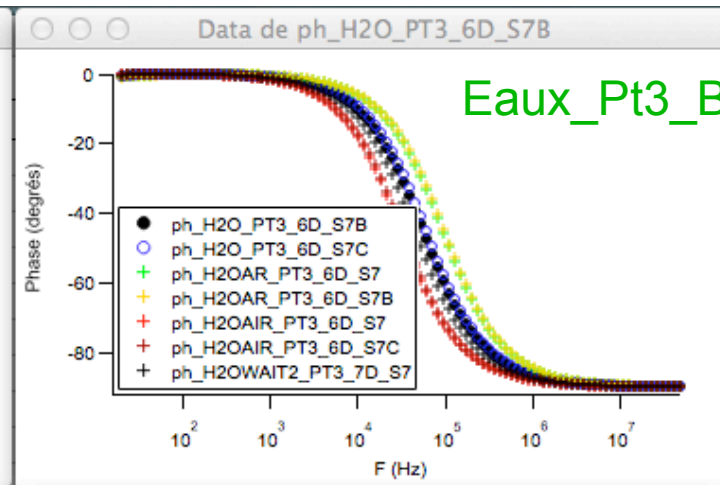
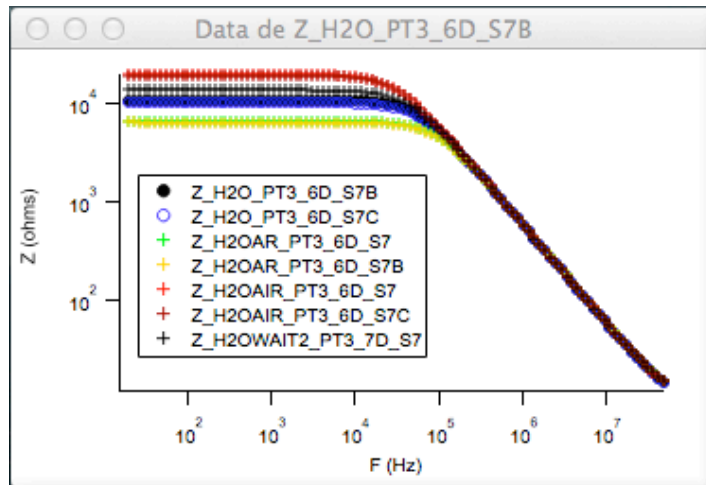


fresh water
time ↑
degassed with argon
30 min later
stayed 5h at air
later.

→ K ↑ with time
Logical.

→ always a step at the beginning
→ step ↓ with time and disappears

Tests water - Cell 3 - $D = 4.1 \text{ mm}$.



H_2O just taken from the machine (Millipore, ultra pure)
 degassed with argon.
 stayed at air several hours
 24 h in a glass bottle closed.

2nd curve: later (15 to 30 min)
 The liquid stays in the cell.

over
 does?

- After 30 min to 2h, the step disappears
- If I wait too long, there is evaporation (leakage in the sandwich cell) $\rightarrow \epsilon \downarrow$ because cell is not full - But this is independent of the step.

